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Chemical Names and  
Formulas. STUDY.  
PLAY. Monatomic ions.  
Consist of a single  
atom with a positive or  
negative charge

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resulting from the loss  
or gain of one or more  
valence electrons,  
respectively. Key 1.

When the metals in  
Group 1A, 2A, and 3A  
lose electrons, they  
form cations with  
positive charges equal  
to their group number.

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## Chemical Names

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Gravity. Created by.

bibiloni1901. Terms in this set (20) Monatomic Ion. Consists of a single atom with a positive or negative charge resulting from the loss of gain of one or more valence electrons and behaves as a unit, yet it still has a ...

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29 terms. hales913.

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periodMs. Steele.

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Monatomic ion. - single  
atom with a positive or  
negative charge  
resulting from the loss  
or gain of electrons,  
respectively.

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## Chemical Names

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Section Review

Empirical formula.

molecular formula.

functional groups.

notation that shows the relative # of atoms in a

molecule. a formula

that shows the lowest

whole # ratio of

elements in a.... a

formula that shows the

true ratio of elements

in a compound.

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Chemistry - Chemical Names and Formulas - Chapter 9. Monatomic ions. Polyatomic ions. Binary compound. Acid. A single atom

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with a positive or negative charge. A tightly bound group of atoms that behaves as a unit and has.... A compound composed of two elements that can be either ionic o....

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Chapter 9 "Chemical Names and Formulas"  
H 2 O 2. Section 9.1  
Naming Ions

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## Chemical Names And Formulas

- OBJECTIVES:
    - Identify the charges on monatomic ions by using the periodic table, and name the ions.
- 3.

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Section 9.4 - Naming  
and Writing Formulas  
for Acids and Bases. An  
acid is a compound

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## Chemical Names

that produces  $H^+$  ions when it dissolves in water. The formula for an acid normally starts with  $H$  and  $O$ . When naming acids, you should first determine whether or not the acid contains oxygen.

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Chapter 9 Chemical Names and Formulas 83  
SECTION 9.3 NAMING AND WRITING FORMULAS FOR

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## Chapter 9

### Chemical Names

MOLECULAR  
COMPOUNDS (pages  
268–270) This section  
explains the rules for  
naming and writing  
formulas for binary  
molecular compounds.  
Naming Binary  
Molecular Compounds  
(pages 268–269)

### **Name Date Class**

### **CHEMICAL NAMES**

### **AND FORMULAS 9**

Naming Ions &  
Counting Atoms:  
Chapter 9, 1 Naming &  
*Page 14/25*

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Chemical Names  
Writing Formulas for  
Ionic Compounds:

Chapter 9.2 Naming &  
Writing Formulas for

Molecular Compounds:

Chapter 9.3 Naming &  
Writing Formulas for

Acids: Chapter 9.4

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vocabulary Answer

Keys Ionic Compound

Naming and Formula

Examples from notes

## **Chapter 9: Chemical Names & Formulas**

Name Date Class

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CHEMICAL NAMES AND FORMULAS 9. Chapter 9 Chemical Names and Formulas 85 Names and Formulas for Bases (page 273) 6. A base is a compound that produces \_\_\_\_\_ when dissolved in water. 7. How are bases named?

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Chemical Names and Formulas 281 CHAPTER  
*Page 16/25*



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Answers

9 Assessment 42. a. 2-  
b. 1+ c. 1- d. 3+ 43. a.  
2+ b. 2+ c. 3+ d. 1+  
44. a. barium ion b.  
iodide ion c. silver ion  
d. mercury(II) ion 45.  
cyanide,  $\text{CN}^-$  and  
hydroxide,  $\text{OH}^-$  46. a.  
hydroxide ion b.  
lead(IV) ion c. sulfate  
ion d. oxide ion 47.  
zero; A compound is  
electrically neutral. 48.  
The symbols for the  
cation and

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### Section Review Answers Chapter 9 Chemical Names and

Formulas 211. Section  
Review. Objectives.

- Determine the charges of monatomic ions by using the periodic table and write the names of the ions.
- Define a polyatomic ion and write the names and formulas of the most common polyatomic ions.

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- Identify the two common endings for the names of most polyatomic ions.

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3:29 PM Page 211  
NAMING IONS 9**

Chapter 9 Chemical  
Names and Formulas  
221 Name Date 4 Write  
formulas for the  
following ionic  
compounds a sodium  
phosphate c sodium  
hydroxide e  
ammonium chloride b

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magnesium d

potassium cyanide 5

Write formulas for  
compounds formed  
from these pairs of ions  
b , 6 Name the  
following compounds ...

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friend; Find other  
activities; ... In samples

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of any chemical compound, the masses of the elements are always in the same proportions any atom or group of atoms that has a negative charge ... place cation name first followed by the anion name: naming polyatomic ...

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Quiz Chemical Names  
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Unit 5: Chemical  
Names & Formulas

Chemical Names &  
Formulas (Chapter 9) 2

3 Monatomic Ion  
Formation &  
Nomenclature

Monatomic ions consist  
of a \_\_\_\_\_ atom with a  
positive or negative  
charge resulting from  
the gain or loss of one  
or more valence \_\_\_\_\_

Cation: any atom or

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Chemical Names and  
Formulas 281 CHAPTER  
9 Assessment 42 a 2- b  
1+ c 1- d 3+ 43 a 2+ b  
2+ c 3+ d 1+ 44 a  
barium ion b iodide ion  
c silver ion d  
mercury(II) ion 45  
cyanide,  $\text{CN}^-$  and  
hydroxide,  $\text{OH}^-$  46 a  
hydroxide ion b  
lead(IV) ion c sulfate  
ion d oxide ion 47 zero;



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A compound is  
electrically neutral 48  
The symbols for the  
cation and

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ecf8427e.