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country, but those are fairly minor quibbles.

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A.P. Chemistry Practice

Test: Ch. 11, Solutions

Name_____ MULTIPLE

CHOICE. Choose the

one alternative that

best completes the

statement or answers

the question. 1)

Formation of solutions

where the process is

endothermic can be

spontaneous provided

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that . A)the
solvent is a gas and
the solute is a solid

**A.P. Chemistry
Practice Test: Ch.
11, Solutions**

MULTIPLE ...

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Chemistry. gases.

liquids. solids.

converting a gas into a
liquid or solid.... -highly
compressible...

-assumes shape of
volume of container ...

-m.... -particles are not

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spread out ... -almost

incompressible..

-assume....

-incompressible and

have definite shape

and volume ...

-molecules....

ap chemistry

chapter 11

Flashcards and

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(Chapter 11) Multiple

Choice (50%) Please

use the following

choices to answer

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questions 1-7. A)

London dispersion

forces B) Ion-ion

attractions C) Dipole-

dipole attractions D)

Dipole-ion attractions

E) Hydrogen bonding

H) Dipole-induced

dipole attractions I) Ion-

induced dipole

attractions ...

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(Chapter 11)

AP Chemistry Chapter

11. Intermolecular

Forces, Liquids, and

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Solids - 10 - Vapor Pressure and Boiling Point • Liquids boil when the external pressure at the liquid surface equals the vapor pressure. • The normal boiling point is the boiling point at 760 mm Hg (1 atm).

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Intermolecular Forces, Liquids ...

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Chapter 11 - Properties
of Solutions . 11.1

Solution Composition .

A. Molarity 1. liters of.
solution moles solute

Molarity(M) = B. Mass

Percent 1. $\times 100 =$

mass of. solution mass
of solute Mass percent.

C. Mole Fraction . 1. D.

Molality 1. kilograms of

solvent moles of solute

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Molality $\log = E.$

Normality 1. liter of
solution equivalents

Chapter 11 - Properties of Solutions

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Key. Key Solutions

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Question . Ch 14 and

19: Ch 14 Ch19

Practice Keys .

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Baker, Mrs. Test
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proportion, and
quantity; structure and
properties of
substances;
transformations; and
energy.

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exam takers, and
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Site Ken MacGillivray -

Hoggard High School ...

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Chapter 3; Chapter 4;

Chapter 5; Chapter 6;

Chapter 7; Chapters 8

& 9; Chapter 10;

Chapter 11; Chapter

12; Chapter 13;

Chapter 14; Chapter

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Solutions. Formula ...

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Properties of
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many AP Chemistry
teachers. You may well
find them useful. They

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Attempts 2 Take the

Quiz.

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Exam: Chemistry 2

AP ...

At a given temperature

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the vapor pressure of pure liquid diethyl ether and isopropyl alcohol are 730 torr and 310 torr, respectively. A solution prepared by mixing isopropyl alcohol and diethyl ether obeys Raoult's law.

**AP Chemistry
Review Questions -
Properties of
Solutions**

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test as without

difficulty as review

them wherever you are

now. Ap Chemistry

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Test AP Chemistry:

Intermolecular Forces,

Liquids, and Solids

Lecture Outline 11.1 A

Molecular Comparison

of Liquids and Solids

Physical properties of

liquids and solids are

due to intermolecular

forces.

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