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Aqueous Systems

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set (15) Surface  
tension. The inward  
force, or pull that tends

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to minimize the surface area of a liquid.

Surfactant. Any substance that interferes with the hydrogen bonding between water molecules and thereby reduces surface tension. Aqueous solution.

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compound that does  
not conduct an electric  
current in aqueous

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Water And  
Aquatic Systems

solution or when  
molten is called a  
what? non-electrolyte.

A system which  
contains particles that  
are 0.1-1 nm in size is  
considered a what?

solution. What are  
some important  
properties of water  
that result from  
hydrogen bonding?

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Causes of Hardness in

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Temporary Hardness  
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Unit-15 Lecture-4**

in the real world  
include rubbing alcohol  
(70% isopropyl alcohol,  
30% water) and  
gasoline (a mixture of  
hydrocarbons). 2.

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Water And Aqueous Systems

solvent, solute 3. When an ionic solute dissolves in water, a given ion is pulled into solution by the attractive ion-dipole force exerted by several water molecules. For example, in

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Check:  $3.15 \times 10^{-3}$   
 $0.050 \times 100\% = 6.28$   
 $\% 3.15 \times 10^{-3} 0.050$   
 $\times 100\% = 6.28\%$  This  
value is greater than  
5%, so a more exact  
method, such as  
successive  
approximations, must  
be used.  $[\text{Ca}^{2+}] = 2.8$   
 $\times 10^{-3} \text{ M}$

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